- (D) (a) What is Hinsberg reagent? How will you distinguish between 1°, 2° and 3° amines by using Hinsberg reagent. 1+3=4
  - (b) Why aniline can not undergo Friedel Craft reaction and nitration reaction?
  - (c) How will you prepare ethylamine by Gabriel synthesis?

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1 (Sem-4) CHE 2

## 2025

## **CHEMISTRY**

Paper: CHE0400204

(Organic Chemistry-I)

Full Marks: 45

Time: 2 hours

## The figures in the margin indicate full marks for the questions.

- 1. Answer the following questions:  $1 \times 5=5$ 
  - (A) Find out the correct answers:
    - (a) Complete hydrolysis of proteins produces—
      - (i)  $NH_3$  and  $CO_2$
      - (ii) Glycogen and fatty acid
      - (iii) Urea and Uric acid
      - (iv) a mixture of amino acids

(b)	Substances which reduce the rate
	of enzyme catalyzed reactions are
	known as:

- (i) substrates
- (ii) enzymes
- (iii) products
- (iv) inhibitors
- (c) The heterocyclic diene employed in cyclo-addition reaction is—
  - (i) Furan
  - (ii) Pyrrole
  - (iii) Thiophene
  - (iv) 2, 5-dimethyl pyrrole

## (B) Fill in the blank:

Hydroxy acids undergo intramolecular esterification in the presence of acid catalyst to yield \_\_\_\_\_.

(C) Write the structure of nicotine.

- 2. Answer the following: (any five)  $2 \times 5 = 10$ 
  - (a) Thiophere is less reactive than furan'. Explain.
  - (b) Why the boiling point of ethylamine  $(CH_3CH_2-NH_2)$  is less than that of ethyl alcohol  $(CH_3CH_2OH)$ .
  - (c) Arrange the following sets of compounds in increasing order of basicity.

    1×2=2

Set I: (i)  $CH_3CH_2 - NH_2$ 

- (ii)  $CH_3CONH_2$
- (iii)  $C_6H_5CONH_2$
- (iv)  $C_6H_5NH_2$

Set II: (i) p-toluidine

- (ii) p-nitroaniline
- (iii) N, N-dimethyl-p-toluidine
- (iv) aniline
- (d) Give the different types of bonds responsible for the tertiary structure of proteins.
- (e) How will you synthesize alanine from ethyl chloride?

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- (f) How can you prepare mono-carboxylic acids from— 1×2=2
  - (i) a primary alcohol
  - (ii) an aldehyde
- (g) Explain briefly why ethyl benzoate (*Ph-COOEt*) can not undergo claisen condensation reaction.
- (h) How will you prepare lactic acid from acetylene?
- (i) Write chemical reactions for the following transformations— 1×2=2

$$(i) CH_2CH_3 COOH$$

$$(ii) \xrightarrow{CH_2OH} \xrightarrow{CH_2COOH}$$

(i) Complete the following reaction—

$$\bigcirc OH + O = N - \bigcirc OH \longrightarrow ? \xrightarrow{NaOH} ?$$

- 3. Answer **any four** from the following:  $5\times4=20$ 
  - (a) (i) How substituted pyridines can be prepared by Hantzsch synthesis? 2
    - (ii) Explain the Fisher Indole synthesis with mechanism.
  - (b) (i) How will you establish the presence of pyridine nucleus in nicotine?
    - (ii) What class of alkaloid does nicotine belongs to?
    - (iii) What happen when aliphatic primary amine is diazotized? 1
    - (iv) Mention one application of diazotization reaction.

- (c) (a) Give one method of preparation of each of— 1×3=3
  - (i) Furan
  - (ii) Pyrrole
  - (i) Thiophene
  - (b) complete the following reactions—
    1×2=2

(i) 
$$\underset{\text{H}}{\boxed{\mid \mid}} \xrightarrow{CHCl_3 ; KOH}$$
 ? 1

(ii) 
$$\bigcap_{N} \xrightarrow{KOH}$$
 ? 1

- (d) (i) Compare the basicities of furan, pyrrole and thiophene. 3
  - (ii) Explain briefly why furan is less reactive than pyrrole. 2
- (e) Describe the following (any two):  $2\frac{1}{2} \times 2=5$ 
  - (i) ISO-electric point of amino acid

- (ii) Denaturation of protein
- (iii) Enzyme inhibitors.
- (f) Write a short note on the effect of ring substituents on the basicities of aromatic amines.
- (g) (i) What is Hoffmann Exhaustive Methylation reaction? 2
  - (ii) Identify the products of the following compounds by using Hoffmann Exhaustive Methylation reaction. 1×3=3

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(h) (i) Explain various types of electronic transmissions possible for organic compounds in *uv*-visible spectroscopy.

OR

- (ii) In IR spectroscopy, absorption signals for molecular vibrations are recorded. What are these molecules vibrations?

  Show the types of molecular vibrations possible in a molecular of the type  $A_2x$  where 'x' is called anchor atom.
- 4. Answer **any four** from the following:  $1 \times 10 = 10$ 
  - (A) Find out the products A and B in the following reactions:  $2 \times 5 = 10$

(i) 
$$CH_3CH_2 - NO_2 \xrightarrow{Br_2/NaOH} A \xrightarrow{Br_2/NaOH} B$$

(ii) 
$$2RCH_2COOR'$$
  $\xrightarrow{(i)} NaOR''$   $A + B$ 

Li

(iii)  $+ CO_2 \longrightarrow A \xrightarrow{H_3O^{\oplus}} R$ 

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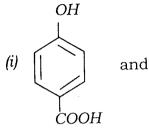
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(iv) 
$$\begin{array}{c} MeCOCl \\ \hline SnCl_4 \\ \hline KOH, CO_2 \\ \hline \end{array}$$
 B

(v) R - CN 
$$\xrightarrow{H^+/H_2O}$$
 A hydrolysis A  $\xrightarrow{H_2/Pd-C}$  Or  $\xrightarrow{LiA|H_4}$ 

- (B) (a) What is chromophore? Give one example.
  - (b) Calculate the  $\lambda_{max}$  of the following compounds—  $2 \times 2=4$

How will you distinguish between— 3



by using IR-spectroscopy

- What is overtone and combination bond?
- Explain why carboxylic acids are (C) much more acidic than alcohol, whereas phenols are weaker acids 3 than carboxylic acids.
  - (b) Find out the correct answer—
  - Which one of the following 5-(I) membered heterocycle is most resonance stabilized?

- Furan (i)
- Thiophene (ii)
- Pyrrole (iii)
- Pyridine (iv)
- In aquous solution, an amino acid (II)exists as-
- cation *(i)*
- anion (ii)

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- dianion
- Zwitter-ion
- Write the name of the optically inactive (c) amino acid.
- Why the electrophilic substitution of (d) furan usually takes place at C-2 position?
- Write Paal-Knorr synthesis of furan. (g)