

- (c) (i) Give the formula, structure and method of preparation of basic beryllium acetate. 1+2+2=5
- (ii) How are poly siloxanes formed? Distinguish between silicon fluids and silicon rubbers. 2+3=5
- (d) Write notes on : **(any two)** 5+5=10
- (i) Pseudohalogens
- (ii) Pasting process
- (iii) Catenation
- (e) (i) State the Pauling's rules for determination of strength of mononuclear oxoacids. 3
- (ii) Arrange the following in order of descending acid strengths in aqueous solution -  
 $\text{HClO}_4$ ,  $\text{HOCl}$ ,  $\text{HClO}_3$ ,  $\text{HClO}_2$   
 Give explanation. 3
- (iii) Pauling's rule is useful in detecting structural anomalies, explain. 2
- (iv) What is symbiosis? Explain. 2
- (f) What is silicates? Explain the bonding and structure of  $\text{SiO}_4^{4-}$  unit using hybridization. What are different types of silicates? Give their structure. 1+4+3+2=10

Total number of printed pages-4

3 (Sem-3/CBCS) CHE HC 1

2023

**CHEMISTRY**

(Honours Core)

Paper : CHE-HC-3016

**(Inorganic Chemistry-II)**

Full Marks : 60

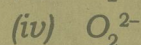
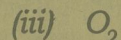
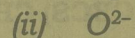
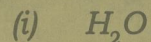
Time : Three hours

**The figures in the margin indicate full marks for the questions.**

- Answer the following as directed : 1×7=7
  - Name the graph of Gibbs Energy ( $\Delta G$ ) versus Temperature (T) for the formation of oxide of metal.
  - "Group-I elements gets oxidized easily" - State whether *True* or *False*.
  - Write the structural formula of borazine.
  - What is "basicity of an acid"?



(e) Which one of the following species is conjugate base of  $\text{OH}^-$ ?



(f) "The name inert gas is improper" - Explain the statement.

(g) Calculate the hardness of  $\text{Al}^{3+}$  for the ionization energy,  $119.99 \text{ eV}$  and electron affinity  $28.45 \text{ eV}$ .

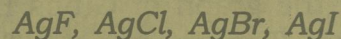
2. Answer the following :  $2 \times 4 = 8$

(a) Describe the structure of boric acid.

(b) What is inert pair effect? Arrange the stability of +1 oxidation states of  $\text{Ca}^+$ ,  $\text{Al}^+$ ,  $\text{In}^+$  and  $\text{Tl}^+$  in their increasing order.

(c) Applying Wade's rule, predict and draw the structure of  $\text{CB}_5\text{H}_9$ .

(d) Arrange the following compounds in ascending order of their solubility in water.



Give explanation.

3. Answer **any three** of the following :

$5 \times 3 = 15$

(a) Briefly discuss the bonding and structure of diborane. 5

(b) What is diagonal relationship? Write *any four* similar properties of  $\text{Be}$  and  $\text{Al}$ .  $1+4=5$

(c) Discuss the Mond's process used in metal refining.

(d) What are polyhalides? Give example. How they are different from Interhalogen Compounds?  $1+1+3=5$

(e) Write constructing properties of the borazine and benzene.

4. Answer **any three** of the following :

$10 \times 3 = 30$

(a) What is Allotropism? Name Different allotropes of carbon. Discuss bonding in graphite. Explain the high thermal and electrical conductivity of graphite. What is intercalation compounds? Give examples.  $2+2+2+2+1+1=10$

(b) (i) What happens when Xenon is heated in presence of flourine? How the amount of flourine affect the nature of product?  $2+2=4$

(ii) Discuss the bonding in  $\text{XeF}_6$ . 4

(iii) Complete the following reaction

